

The Governing Equations

(Rectangular Coordinates)

Velocity Conservation Equations (Equations of Motion) (non-rotating)

$$\frac{Du}{Dt} = -\frac{1}{\rho} \frac{\partial p}{\partial x} + \frac{\mu}{\rho} \left(\frac{\partial^2 u}{\partial y^2} + \frac{\partial^2 u}{\partial z^2} \right)$$

$$\frac{Dv}{Dt} = -\frac{1}{\rho} \frac{\partial p}{\partial y} + \frac{\mu}{\rho} \left(\frac{\partial^2 v}{\partial x^2} + \frac{\partial^2 v}{\partial z^2} \right)$$

$$\frac{Dw}{Dt} = -\frac{1}{\rho} \frac{\partial p}{\partial z} - g + \frac{\mu}{\rho} \left(\frac{\partial^2 w}{\partial x^2} + \frac{\partial^2 w}{\partial y^2} \right)$$

Temperature Conservation Equation

$$\frac{DT}{Dt} = \text{rate of change due to radiative absorption}$$

– rate of change due to radiative emission

+/- rate of change due to phase changes of water

+/- rate of change due to conduction

+/- rate of change due to compression/expansion
associated with changes in pressure ($\propto Dp/Dt$)

Density Conservation Equation (Continuity Equation)

$$\frac{D\rho}{Dt} = -\rho \left(\frac{\partial u}{\partial x} + \frac{\partial v}{\partial y} + \frac{\partial w}{\partial z} \right)$$

Equation of State

- For gases (Ideal Gas Law): $p = \rho RT$
- For liquids: Complex empirical relationship among pressure, temperature, density, and composition of the liquid (careful lab observations fitted to a set of polynomial curves)

**The Governing Equations
in Tendency Form
(Rectangular Coordinates)**

Velocity Tendency Equations

$$\frac{\partial u}{\partial t} = -u \frac{\partial u}{\partial x} - v \frac{\partial u}{\partial y} - w \frac{\partial u}{\partial z} - \frac{1}{\rho} \frac{\partial p}{\partial x} + \frac{\mu}{\rho} \left(\frac{\partial^2 u}{\partial y^2} + \frac{\partial^2 u}{\partial z^2} \right)$$

$$\frac{\partial v}{\partial t} = -u \frac{\partial v}{\partial x} - v \frac{\partial v}{\partial y} - w \frac{\partial v}{\partial z} - \frac{1}{\rho} \frac{\partial p}{\partial y} + \frac{\mu}{\rho} \left(\frac{\partial^2 v}{\partial x^2} + \frac{\partial^2 v}{\partial z^2} \right)$$

$$\frac{\partial w}{\partial t} = -u \frac{\partial w}{\partial x} - v \frac{\partial w}{\partial y} - w \frac{\partial w}{\partial z} - \frac{1}{\rho} \frac{\partial p}{\partial z} - g + \frac{\mu}{\rho} \left(\frac{\partial^2 w}{\partial x^2} + \frac{\partial^2 w}{\partial y^2} \right)$$

Temperature Tendency Equation

$$\frac{\partial T}{\partial t} = -u \frac{\partial T}{\partial x} - v \frac{\partial T}{\partial y} - w \frac{\partial T}{\partial z} + \text{rate of change due to radiative absorption}$$

– rate of change due to radiative emission

+/- rate of change due to phase changes of water

+/- rate of change due to conduction

+/- rate of change due to compression/expansion
associated with changes in pressure ($\propto Dp/Dt$)

Density Tendency Equation (Continuity Equation in Tendency Form)

$$\frac{\partial \rho}{\partial t} = -u \frac{\partial \rho}{\partial x} - v \frac{\partial \rho}{\partial y} - w \frac{\partial \rho}{\partial z} - \rho \left(\frac{\partial u}{\partial x} + \frac{\partial v}{\partial y} + \frac{\partial w}{\partial z} \right) \text{ (advective form)}$$

$$= - \left(\frac{\partial \rho u}{\partial x} + \frac{\partial \rho v}{\partial y} + \frac{\partial \rho w}{\partial z} \right) \text{ (mass flux divergence form)}$$

Equation of State

- For gases (Ideal Gas Law): $p = \rho RT$
- For liquids: Complex empirical relationship among pressure, temperature, density, and composition of the liquid (careful lab observations fitted to a set of polynomial curves)